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A HIGHLY SENSITIVE METHOD TO IDENTIFY THE D,L-CONFIGURATIONS OF MONOSACCHARIDES BASED ON (-)-TBMB CARBOXYLIC ACID AND HPLC

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ABSTRACT

D,L-Isomers of monosaccharides were coupled with a fluorescent chiral derivatizing agent, (-)-2-*tert*-butyl-2-methyl-1,3-benzodioxole-4-carboxylic acid [(-)-TBMB carboxylic acid], to afford a pair of diastereoisomers. The two were well separated with HPLC (reverse phase conditions; ODS, $CH_3CN:H_2O:isoPrOH$) within 60 min and could be detected by fluorescence of the TBMB group at the few picomolar level.

INTRODUCTION

Chemical or enzymic syntheses of sugars with unnatural L-configuration are of interest to us because these sugars may have different biological properties from the corresponding D-isomers, for example in their interactions with lectins or antibodies. Moreover, some monosaccharides like galactose, fucose, arabinose and rhamnose are found as naturally occurring L-sugars. Thus, identification of the D,L-configuration should be important when these sugars are isolated in nature. This is, however,

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difficult especially when the amount of isolated sugars is not enough for the measurements of optical rotations nor enzymic degradations. Development of more sensitive methods to identify the D and L-isomers, therefore, are important.

Several methods have been developed¹⁻⁸ after the gas chromatographic method of Vliegenthart *et al.*.^{3,4} Recently, we reported an ¹H NMR method⁸ using a chiral derivatizing agent, (-)-TBMB carboxylic acid⁹⁻¹¹ (SCHEME) in which strong signals from the *tert*-butyl and methyl groups of the derivatized sugar provided diagnostic signals for discriminating D-sugars from L-sugars. In the present paper, we would like to show that D,L-monosaccharides can be separated by HPLC after derivatization with this agent to form a pair of diastereoisomers. Taking advantage of TBMB carboxylic acid as a fluorescence marker will enable us to detect and discriminate between D,L-isomers of monosaccharides in a highly sensitive manner.

RESULTS AND DISCUSSIONS

In our previous NMR study,⁸ we reported a general route to derivatize monosaccharides with (-)-TBMB carboxylic acid (SCHEME) Here, the derivatization procedure was modified for micro analysis by HPLC. The typical procedure is as follows; reducing sugar (ca. 0.01 mg, ca. 50 nmol) was treated at room temperature with 0.1 mL of acetic anhydride containing a catalytic amount of $HClO_4$ (ca. 0.01 %). After 15 min, 0.1 mL of EtOH was added, and the mixture was kept at room temperature for 30 min. K₂CO₃ (1 mg) and 0.1 mL of CH₂Cl₂ were added, and the mixture was centrifugated. The supernatant was collected and concentrated in vacuo. The residue was dissolved in 0.1 mL CH₂Cl₂ and treated with 0.2 mL of 33% HBr/AcOH (commercially available) for 1h. The mixture was concentrated in vacuo below 60 $^{\circ}$ C and dissolved in acetone (0.1 mL). To the solution was added an excess amount of (-)-TBMB carboxylic acid and an equimolar amount of potassium hydrogern carbonate. The mixture was warmed at 60 °C for 1 h. A portion of the solution was applied to silica gel TLC coated on a aluminium sheet (5 cm x 5 cm, available as DC-Alufolien from Merck) and developed with toluene-ethyl acetate (2/1).



FIGURE. HPLC separations of D,L-isomers of rhamnose (a), glucose (b), galactose (c) and fucose (d). (HPLC conditions are cited under the Table).

Sugars	Rt(min)	Configu- rations	α°	Rsb
Xylose	28.3 29.8	L (1R) D (1S)	1.05	1.39
Arabinose	24.8 25.7	D (1R) L (1S)	1.04	0.93
Glucose	28.5 30.1	L (1R) D (1S)	1.06	1.33
Galactose	27.5 29.3	L (1R) D (1S)	1.07	1.69
Fucose	30.6 32.4	L (1R) D (1S)	1.06	1.32
Mannose (1,2- <i>trans</i>)	27.1 28.5	D (1R) L (1S)	1.04	1.05
Mannose (1,2-cis)	22.3 22.8	L (1R) D (1S)	1.02	0.45
Rhamnose (1,2-trans)	36.9 40.3	D (1R) L (1S)	1.09	2.00
Rhamnose (1,2-cis)	30.0 30.0	L (1R) D (1S)	1.00	0.00

Table Reverse Phase HPLC Separations of 1-O-(-)-TBMB-D- and L-Sugars.

Condition: ODS column (150 mm x 4.6 mm ϕ), solvents: CH₃CN:H₂O:isPrOH = 4:4:1, Flow rate = 0.8 mL/min.

a. Separation factor = (retention volume of one diastereoisomer - void volume of the column)/(retention volume of another diastereoisomer - void volume of the column). b. Resolution factor = $2 \times (\text{distance between the peaks of two diastereoisomers})/(\text{sum of band width of the two peaks}).$ A fluorescent band (Rf = ca. 0.5) was cut off from the sheet, the silica gel extracted with CH₃CN, and a sample from the extract injected onto an HPLC column (ODS column, 4.6 mm ϕ x 15 cm; CH₃CN:H₂O:*i*soPrOH = 4:4:1) for analysis.

The coupling reaction yielded 1,2-*trans* pyranosides exclusively for Glc, Gal, Fuc, Xyl and Ara. Man and Rha with an *axial* OAc-2 group gave a mixture of 1,2-*trans* and *cis* pyranosides in *ca*. 2:1 ratio. The isomers could be easily separated on silica gel TLC at the final stage. When racemic TBMB carboxylic acid and D-glucose were used for the reaction, a 1:1 mixture of the two diastereoisomers of 1-*O*-[(+)- and (-)-TBMB carbonyl]-D-glucose derivatives were obtained. This means that there is no D,L-discrimination by the chirality of (-)-TBMB carboxylic acid in the present coupling reaction.

HPLC results in the FIGURE and from the summarized data in the Table indicated that D,L-isomers of TBMB carbonylated sugars were well separated in 20-45 min under the conditions described above. Although the D,L-separation of 1,2-*cis* anomers of mannose and rhamnose was insufficient, the main 1,2-*trans* anomers were well separated enough to carry out the D,L-sugar identification.

As expected, the elution order of the D,L-isomers was governed by the anomeric configuration; (1R)-isomer was always eluted faster than the (1S)-isomer (Table), and this rule is practically useful for the assignment of the D,L-configuration since the coupling reaction affords mainly 1,2-trans isomers.

CONCLUSIONS

An HPLC method was developed to determine the D,L-configurations of monosaccharides which involves a coupling reaction with (-)-TBMB carboxylic acid and per-O-acetyl pyranosyl bromides to give diastereoisomeric 1-O-TBMB carbonylated sugars which could be separated by HPLC. The minimum amount of sugars for the fluorescence detection could be reduced to a few pico mole using an HPLC column, thus allowing highly sensitive discrimination between the D,L-sugar isomers. This approach will become a complementary way to the NMR spectroscopy method reported previously.⁸

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